The massive star is closer to center of mass, and moves more slowly than the planet, but it does move!

Worksheet \#5: Sun and Jupiter orbit their common center of mass every 11.86 years.
Note M_J / M_Sun = 1/1047
What is the orbital speed of the Sun?
What is the astrometric displacement for a similar system at a distance of 10 pc ?

Solution: First calculate the semi-major axis of the Jupiter's orbit $\mathrm{P}^{2}=\mathrm{a}^{3}$ so $\mathrm{a}=5.2 \mathrm{AU}=780 \times 10^{9} \mathrm{~m}$
$M_{s} r_{s}=M_{j} r_{j}$ so $r_{s}=740 \times 10^{6} \mathrm{~m}$
Angular size $=2 \times 740 \times 10^{6} \mathrm{~m} / 10 \times 3.09 \times 10^{16} \mathrm{~m}$
$=4.8 \times 10^{-9}$ radians $\times 206265000 \mathrm{mas} / \mathrm{rad}$
$=0.99 \mathrm{mas}$

Calculate orbital speed of Sun assuming Jupiter is only planet:
Moves in circular orbit of radius $742,000 \mathrm{~km}$

How much Doppler shift? Consider H-alpha absorption line, at rest wavelength 656 nm :

$$
V=\frac{2 \pi r}{P}=\frac{2 \pi(742,000 \mathrm{~km})}{11.86 \text { years }}=12.5 \mathrm{~m} / \mathrm{s}
$$

$\Delta \lambda=\frac{V}{c} \lambda_{0}=\frac{12.5 \mathrm{~m} / \mathrm{s}}{3 \times 10^{8} \mathrm{~m} / \mathrm{s}} 656 \mathrm{~nm}=2.7 \times 10^{-5} \mathrm{~nm}$

## 51 Pegasi

Michel Mayor \& Didier Queloz at Geneva Observatory observed wobble in 1995
Sun-like star 15 pc distance


Wobble was $53 \mathrm{~m} / \mathrm{s}$, period 4.15 days

Implied a planet with 0.5
Jupiter mass orbiting at 0.05 AU!
First planet found around sunlike star


Michel Mayor \& Didier Queloz Nobel Prize 2019

## Selection effects

Doppler wobble biased towards massive planets close to their star (leads to larger velocities and shorter periods). Now getting close to Earth-mass planets.

Limited by the orbital speed sensitivity (few m/s but always improving) and length of orbital period: for more than several year periods, hard to tell if motion is periodic

Inclination of binary orbit unknown (unless transits observed). More likely to be close to edge-on for detection. If not, wobble is larger than measured and so is planet.

## Characteristics of detections

Some hot Jupiters on small orbits (migration or formation?)

Some with very elliptical orbits (in the Solar System this is a sign of perturbation $=>$ supports migration idea?)

Rare around low-mass stars (smaller disks thus less material)

Rare around metal-poor stars

count: 4472 planets! What about these "hot Jupiters"?


Planetary Mass (Mjup)




## TRAPPIST-1 System



## Kepler-186 f <br> Earth

HABITABLE ZONES OF PLANETARY SYSTEMS
Planets to scale with one another but not to orbital distances.



## Detecting Radio Bursts from Exoplanets

Suitability of the LWA1
Observations to date
Near future: Owens Valley
Farther future: the LWA swarm


## Emission from Jupiter



## Also, sort the following table out on your own from the physics and geometry involved in each type of event:

| Type of binary | Observations performed (or needed) | Parameters determined |
| :---: | :---: | :---: |
| Visual | a) Apparent magnitudes and $\pi$ <br> b) P, a, and $\pi$ <br> c) Motion relative to CM | Stellar luminosities <br> Semi-major axis (a) <br> Mass sum (M+m) |
| Spectroscopic | a) Single velocity curve <br> b) Double velocity curve | Mass function $\mathrm{f}(M, m)$ Mass ratio (M/m) |
| Eclipsing | a) Shape of light curve eclipses <br> b) Relative times between eclipses <br> c) Light loss at eclipse minima | Orbital inclination (i) <br> Relative stellar radii ( $\left.R_{l, s} / a\right)$ <br> Orbital eccentricity (e) <br> Surface temperature ratio $\left(T / T_{s}\right)$ |
| Eclipsing/spectroscopic | a) Light and velocity curves <br> b) Spectroscopic parallax + apparent magnitude | Absolute dimensions ( $a, r_{s}, r_{l}$ ) $e$ and $i$ <br> Distance to binary <br> Stellar luminosities <br> Surface temperatures ( $T_{l}, T_{s}$ ) |

## Astronomy 421



Lecture 8: Stellar Spectra

## Key concepts:

## Stellar Spectra

The Maxwell-Boltzmann Distribution

The Boltzmann Equation

The Saha Equation

## UVBRI system

| Filter name | Effective wavelength <br> $(\mathrm{nm})$ | 0-magnitude flux (Jy) |
| :--- | :--- | :--- |
| U | 360 | 1880 |
| B | 440 | 4400 |
| V | 550 | 3880 |
| R | 700 | 3010 |
| I | 880 | 2430 |



Color index useful since it defines a star's temperature


## Stellar Spectra



## Stellar spectral types

| Spectral type | Temperature of Atmosphere | Examples |
| :--- | :--- | :--- |
| O | $30,000-50,000$ |  |
| B | $10,000-30,000$ | Rigel |
| A | $8,000-10,000$ | Vega, Sirius |
| F | $6,000-8,000$ |  |
| G | $5200-6000$ | Sun |
| K | $4000-5200$ |  |
| M | $2000-4000$ | Betelgeuse |
| L | $1300-2000$ |  |
| T | $<1300$ |  |

Further subdivision: e.g., B0-B9, G0-G9 etc. The Sun is a G2.



Why is spectrum so sensitive to temperature (apart from blackbody)?

Schematic of stellar atmosphere





Atmosphere: atoms, ions, electrons


Surface


## Maxwell-Boltzmann velocity distribution

Consider gas in a star with some density, temperature, chemical composition (may change with $R$ ).

Several different atomic processes occur:

- Collisional excitation
- Collisional de-excitation
- Radiative excitation
- Radiative de-excitation
- Collisional ionization
- Radiative ionization
- Recombination

Thermal Equilibrium: $T$ of any parcel of gas equals that of surroundings. Not quite true in stars, but $T$ varies slowly enough with $R$ that it's a good approximation.

In TE, the number density of particles with speed between $v$ and $v+d v$ :

$$
n_{V} d v=n\left(\frac{m}{2 \pi k T}\right)^{3 / 2} e^{-m v^{2} / 2 k T} 4 \pi v^{2} d v
$$

This is the Maxwell-Boltzmann velocity distribution function. $k T$ is the characteristic thermal energy of a gas at temperature T .

Most particles have KE~kT, due to collisions. This process is called thermalization.

This is a result from "statistical mechanics".


Most probable speed $\quad v_{m p}=\sqrt{\frac{2 k T}{m}}$

Root mean-square speed $\quad v_{r m s}=\sqrt{\frac{3 k T}{m}}$

Note: the speed depends on mass. At a given $T$, more massive particles slower on average, but each species has same average KE.

Point: Atoms of a gas gain and lose energy via collisions.
In typical stellar atmospheres, radiative transitions dominate.

## Maxwell-Boltzmann Velocity Distribution for

 $\mathrm{N}_{2}$

## What dictates spectral line strengths?

We'll see spectral lines are extremely useful - relative strengths give information on temperature (most clearly), density, and composition of stellar atmospheres.

Consider the case of thermal equilibrium (TE) => average number of atoms in a given energy (i.e. electron energies) state remains unchanged, e.g. each excitation balanced by a de-excitation ("steadystate").

The relative number of atoms or ions in each state is governed by the Boltzmann Equation:

$$
\frac{N_{b}}{N_{a}}=\frac{g_{b} e^{-E_{b} / k T}}{g_{a} e^{-E_{a} / k T}}=\frac{g_{b}}{g_{a}} e^{-\left(E_{b}-E_{a}\right) / k T}
$$

$$
\frac{N_{b}}{N_{a}}=\frac{g_{b} e^{-E_{b} / k T}}{g_{a} e^{-E_{a} / k T}}=\frac{g_{b}}{g_{a}} e^{-\left(E_{b}-E_{a}\right) / k T}
$$

$N_{b}=$ \#atoms per unit volume in state $b$
$N_{a}=$ \#atoms per unit volume in state $a$
$E_{b}=$ energy of level $b$
$E_{a}=$ energy of level a
$k=$ Boltzmann's constant $=1.38 \times 10^{-23} \mathrm{~J} \mathrm{~K}^{-1}=8.6 \times 10^{-5} \mathrm{eV} \mathrm{K}^{-1}$
$T=$ gas temperature
$g_{i}=$ multiplicity of level $i$
= how many $e^{-}$you can put in level $i$ before Pauli exclusion principle is violated
= "statistical weight" or "degeneracy" of energy level $i$ (number of quantum states (different $I, m_{l}, m_{s}$ ) with same energy)

Statistical weight for H atom
$n=1, l=0, m_{l}=0, m_{s}=+/-1 / 2 \Rightarrow g=2$
$n=2, l=0,1, \quad m_{l}=0,+/-1, \quad m_{s}=+/-1 / 2 \Rightarrow \quad g=8$

Can show $g_{n}=2 n^{2}$ for H .

Boltzmann Equation qualitatively:

$$
\frac{N_{b}}{N_{a}}=\frac{g_{b}}{g_{a}} e^{-\left(E_{b}-E_{a}\right) / k T}
$$

Physical reasoning:

- $\frac{N_{b}}{N_{a}} \uparrow$ as $T \uparrow \quad$ more collisional, radiative excitations at higher $T$
- $\frac{N_{b}}{N_{a}} \rightarrow \frac{g_{b}}{g_{a}}$ as $T \rightarrow \infty \quad$ exponential decline becomes unimportant
- $\frac{N_{b}}{N_{a}}$ small if $E_{b}-E_{a} \gg k T$

Few excitations to level $b$ if typical thermal energy too low.

## Example

What is $N_{2} / N_{1}$ for hydrogen for the Sun? $T \sim 5780 \mathrm{~K}$.

$$
\begin{aligned}
E_{n} & =-\frac{13.6 e V}{n^{2}} \\
\frac{N_{2}}{N_{1}} & =\frac{g_{2}}{g_{1}} e^{-\left(E_{2}-E_{1}\right) / k T} \\
& =\frac{2(2)^{2}}{2(1)^{2}} e^{-\left[\left(-13.6 \mathrm{eV} / 2^{2}\right)-\left(-13.6 \mathrm{eV} / 1^{2}\right)\right] / k T}=4 e^{-10.2 \mathrm{eV} / k T} \simeq 4 \times 10^{-9}
\end{aligned}
$$

Balmer lines are not very intense in the Solar spectrum.

For stars T~10,000 K, $\frac{N_{2}}{N_{1}} \sim 3 \times 10^{-5}$
So stronger Balmer lines in Type A stars.
In fact, to reach $N_{b} / N_{a}=1$, we need $T=85,000 \mathrm{~K}$.


Observationally though, Balmer lines reach maximum intensity at $\mathrm{T}=9250 \mathrm{~K}$, and fall in intensity at higher temperatures. Why?

Ionization!

As $T$ increases, there is more energy (both radiative and collisional) available to ionize the atoms.

In equilibrium, the ionization rate = recombination rate for every type of ion.

$$
X \leftrightarrow X^{+}+e^{-}
$$

The ratio of the number of atoms in ionization stage (i+1) to the number in stage $i$ :

$$
\frac{N_{i+1}}{N_{i}}=\frac{2 Z_{i+1}}{n_{e} Z_{i}}\left(\frac{2 \pi m_{e} k T}{h^{2}}\right)^{3 / 2} e^{-\chi_{i} / k T}
$$

The Saha Equation

Where
$\chi_{\mathrm{i}}=$ ionization energy needed to remove $e^{-}$from atom in ground state of state $i$ to $i+1$
$n_{e}=$ free electron number density
$Z=$ partition function, weighted sum of ways the atom or ion can distribute its electrons among its energy levels

$$
Z=\sum_{j=1}^{\infty} g_{j} e^{-\left(E_{j}-E_{1}\right) / k T}
$$

weighted such that higher $E$ is a less likely configuration.
Accounts for fact that not all atoms or ions will be in ground state. $Z$ must be calculated for each ionization state of the element.

For an ideal gas, can also express Saha Eqn in terms of pressure, using $P_{e}=n_{e} k T$

$$
\frac{N_{i+1}}{N_{i}}=\frac{2 k T Z_{i+1}}{P_{e} Z_{i}}\left(\frac{2 \pi m_{e} k T}{h^{2}}\right)^{3 / 2} e^{-\chi_{i} / k T}
$$

Example: consider a pure Hydrogen atmosphere at constant pressure:

$$
P_{e}=20 \mathrm{Nm}^{-2}
$$

We want to calculate the ionized fraction as a function of $T$, from 5000 K up to $25,000 \mathrm{~K}$.

Then we need the partition functions $Z_{l}$ (neutral) and $Z_{/ /}$(ionized).
$Z_{\text {II }}=1 \quad$ - only one state available, just a proton.
$Z_{l}$ : at these $T \mathrm{~s}, E_{2}-E_{1}=10.2 \mathrm{eV} \gg k T$. (check!). Thus, as before, most neutral H is in ground state, and only $\mathrm{j}=1$ contributes significantly to $Z_{i}$ :

$$
Z_{I} \simeq g_{1}=2(1)^{2}=2
$$

Plug this into the Saha Equation giving $N_{I I} / N_{I}$ and then compute the fraction of ionized hydrogen which must be $N_{I I} / N_{t o t}$

$$
\frac{N_{I I}}{N_{t o t}}=\frac{N_{I I}}{N_{I}+N_{I I}}=\frac{N_{I I} / N_{I}}{1+N_{I I} / N_{I}}
$$



So, ionization happens in a very narrow range of $T$. Almost completely ionized (95\%) by $T=11,000 \mathrm{~K}$.

Combine Boltzmann and Saha Equations to understand the Balmer lines:
Population of higher levels with higher $T$ (Boltzmann) quenched by ionization (Saha)

Therefore, at some $T$, the population of $e^{-}$in higher levels will reach a maximum.
For our hydrogen example, at $\mathrm{T}<25,000 \mathrm{~K}$, almost all neutral atoms in $n=1$ or 2 , so $N_{I} \approx N_{1}+N_{2}$ and population of $n=2$ level is:

$$
\frac{N_{2}}{N_{t o t}}=\frac{N_{2}}{N_{2}+N_{1}} \frac{N_{I}}{N_{t o t}}=\frac{N_{2} / N_{1}}{1+N_{2} / N_{1}} \frac{1}{N_{t o t} / N_{I}}
$$

since $N_{t o t}=N_{I}+N_{I I}$

$$
\frac{N_{2}}{N_{t o t}}=\frac{N_{2} / N_{1}}{1+N_{2} / N_{1}} \frac{1}{\left(1+N_{I I} / N_{I}\right)}
$$

## Plot this


$n=2$ maximized at $T \sim 10,000 \mathrm{~K}$ (A stars) => Balmer absorption lines $\left(n_{\text {lower }}=2\right)$ strongest.

## Why are lines of other elements often at least as strong as $\mathbf{H}$ ?

For example, Ca in the Sun has $\frac{N_{C a}}{N_{H}}=2 \times 10^{-6} \quad$ !
But $\mathrm{CaII}=>$ Ca II requires only 6.11 eV . So $\sim$ all Ca is Ca II.

Ca II "H" and "K" lines at $\sim 400 \mathrm{~nm}$ require only 3.12 eV photons. Absorption is from ground state.

Balmer lines require electrons to be in $n=2,10.2 \mathrm{eV}$ above $n=1$.

Because of exponentials in Boltzmann and Saha equations, there are many more Ca II's in ground state than H's in $n=2$.

> (see example 8.1.5 for numbers)

Temperature (K)


## Hertzsprung Russell Diagram



More commonly made from observations: a color-magnitude diagram:

(globular cluster M13)

## (Morgan-Keenan) Luminosity Classes

Atmospheric pressure affects width of absorption lines:
Lower pressure => decreased line width
Higher pressure => increased line width

The atmospheric pressure is lower in the photosphere of an extremely large red giant than in a main sequence star of similar temperature => giant's spectral lines are narrower


Sun G2V Betelgeuse M2la

Cartoon of a Hertzsprung-Russell Diagram with Luminosity Classes


With spectral type and luminosity class, distance to star can be estimated: "spectroscopic parallax".

