ASTR2115 Angles, Units, Temperature, Light

Chapters 1 and 5

Announcements

• First Homework is due next week

The process of astronomy research

- Build a telescope
- What is it that we see?
- How does it work?

- (instrumentation)
 - (observing)
 - (analysis)
- How was it formed, and how will it evolve? (theory & predictions)

Example of angular distance: the "pointer stars" in the big dipper



The Moon subtends about one-half a degree

Angular size - linear size - distance

Use the *small-angle formula*:

$$D = \frac{\alpha d}{206265}$$

where D = linear size of an object (any unit of length), d = distance to the object (*same* unit as D) α = angular size of the object (in arcsec),

The 206,265 is required in the formula – it's the number of arcseconds in a circle divided by 2π .

Where does this formula come from?

The same idea in pictures: the angular size depends on the linear (true) size AND on the distance to the object. See Box 1-1.



Moving an object farther away reduces its angular size.

 $C = 2\pi r$

As long as the angular size is small, we can think of the object's physical size as a small piece of a circle.

Units in astronomy

Astronomers use the normal metric system and powers-of-ten notation, plus a few "special" units.

Example: Average distance from Earth to Sun is about 1.5×10^{11} m = 1 Astronomical Unit = 1 AU

Used for distances in the Solar system.

This fall we are working on much larger scales, in which case we normally use the unit of parsec.

The parsec unit

- Relation between angular size and linear distance => basic unit of distance in astronomy: the parsec
- Short for "parallax of one second of arc"
- The distance between Earth and a star at which the radius of the Earth's orbit around the Sun (1AU) subtends an angle of 1"

The nearest star is 1.3 pc away (1 pc = $3.09 \times 10^{16} \text{ m}$)



Positions of celestial objects

- Same idea when we describe the position of a celestial object
- The Sun, the Moon and the stars are so far away that we cannot perceive their distances by eye no depth perception.
- Instead, the objects appear to be projected onto a giant, imaginary sphere centered on the Earth.
- All objects seem to be on the surface of this imaginary sphere

The celestial sphere

- Stars are really distributed in 3-dimensional space, but they *look* like they' re on a 2-dimensional sphere centered on Earth.
- To locate an object, two numbers (angular measures), like longitude and latitude are sufficient.
- Useful if we want to decide where to point our telescopes.



Horizon coordinate system

- The Horizon coordinate system (used for telescopes):
- Altitude
 - Angle above the horizon
 - 0-90°
 - The altitude of the north celestial pole equals the observer's latitude on Earth.



• Azimuth

- Angle measured eastward along horizon, starting from the north
- 0-360°

Pros and cons of the Horizon system

Pros

Easy to tell and understand

Cons

- At different position on the Earth, the same object has different coordinates
- At different times, the same object has different coordinates

The coordinates of an object change in this system!



The Equatorial system

- A system in which the coordinates of an object does NOT change.
- The coordinates are called Right Ascension and Declination, and are analogous to longitude and latitude on Earth.



• The equatorial coordinate system rotates with stars and galaxies.

RA and Dec

- Declination (Dec) is measured in degrees, arcminutes, and arcseconds.
- Right ascension (RA) is measured in units of time: hours, minutes, and seconds.
- Example 1: The star Regulus has coordinates RA = 10^h 08^m 22.2^s Dec = 11° 58' 02"

Stellarium Demo

Zero point of RA: *The vernal equinox,* which is the point on the celestial equator the Sun crosses on its march north - the start of spring in the northern hemisphere (cf. Greenwhich 0° longitude).

Properties of Light Chapter 5



Light carries information to astronomers

- How hot is the Sun?
- How does it compare to other stars?
- What is its composition?
- Is there anything between the stars? Between galaxies?
- How do we know that the Universe is expanding?

All of this comes from the study of light!

Astronomers use telescopes to collect light of distant objects, and then extract information from this light.

What is light?

- Light is *electromagnetic (EM) radiation*
- Light can be treated either as
 - EM waves
 - Photons (particles of light)
- Both natures have to be considered to describe all essential properties of light

Light as electromagnetic waves

- EM waves: self propagating electric and magnetic fields (changes in strengths of E- and B-fields).
- Traveling (in vacuum) at the constant speed of light $c = 3 \times 10^8$ m/s.
- $v = c/\lambda$, where v is the frequency [Hz] and λ is the wavelength [m].



EM waves are different from other waves, since it doesn't need a medium to propagate in!

Properties of light

- It heats up material (e.g. your skin), thus it is a carrier of energy.
 - This is radiative energy, one out of three basic categories of energy (kinetic, potential and radiative)
- It has a "color", especially in the visible regime you are used to the colors of the rainbow:



 The human eye is sensitive to light with colors from violet to red, which corresponds to a wavelength range:

4,000 Å < λ < 7,000 Å

= 400 nm < λ < 700 nm where 1 nm =10⁻⁹ m.

where an Å is 10^{-10} m (Å=Ångström).



- White light: all colors of the rainbow
- Black: no light, hence no colors

The electromagnetic spectrum

 Light extends beyond the visible regime, and the continuous and infinite distribution of wavelengths is called the electromagnetic spectrum



Light of different wavelengths interacts in different ways with matter

The Radio Astronomy Group Department of Physics and Astronomy



The electromagnetic spectrum



The dual nature of light

Light often acts like a wave but also can act like a particle – in interactions with atoms and molecules.

- The particles are called *photons*.
- A photon is a mass-less particle that carries energy E at the speed of light.
- $E = hc/\lambda = hv$ where $h = 6.6 \times 10^{-34} \text{ J s}$ (Planck's constant)

Question: which has more energy, a photon of blue light, or of red light? UV or radio?

Young's double slit experiment

Demonstrates the wave-like nature of light: Like water waves (right), EM waves can interfere.



The photoelectric effect

- Illustrates the particle nature of light: Photons hitting a piece of metal will knock out single electrons, but only if they have enough energy.
- Above this energy level the kinetic energy of the photon does NOT depend on the intensity of the incident radiation, but on the frequency.



 IF light had only wave-nature, the kinetic energy of ejected electrons should depend on the amplitude (intensity) of the wave, but instead it depends on the frequency (E=h_V)

How do light and matter interact?

- Emission light bulb, star
- Absorption your skin can absorb light, in turn the absorbed energy heats your skin
- Transmission glass and air lets the light pass through
- Reflection and scattering light can bounce off matter leading to reflection (same direction of reflected light) or scattering (random direction of reflected light)

Materials that transmit light are *transparent*. Materials that absorb light are *opaque*.

Three basic types of spectra

- Kirchoff's laws of spectroscopy:
 - 1. A hot, opaque body, or a hot, dense gas produces a continuous spectrum.
 - 2. A hot, transparent gas produces an emission line spectrum.
 - 3. A cool, transparent gas in front of a source of a continuous spectrum produces an absorption line spectrum.



Illustration Kirchoff's laws



Astrophysical examples:

- Continuous: asteroids, planets, etc.
- Emission line: hot interstellar gas -- HII regions, planetary nebulae, supernova remnants.
- Absorption line: stars (relatively cool atmospheres overlying hot interiors), cool interstellar gas.

Important concepts: temperature and blackbody.

Kelvin temperature scale

- An absolute temperature system in which the temperature is directly proportional to the internal energy.
 - Uses the Celsius degree, but a different zero point.
 - 0 K: absolute zero
 - 273 K: freezing point of water
 - 373 K: when water boils

Thermal radiation

- Let's consider the concept of a "blackbody"
 - Tenuous gas: most light can pass through relatively unaffected
 - Denser matter (like a rock) will not allow light to pass as easily
- A dense body will absorb light photons over a wide range of wavelengths
- This also means the absorbed photon cannot easily escape once reemitted it immediately interacts with another molecule
- The photons bounce around inside the body for a long time, and when they finally escapes it has randomized wavelength

=> results in a smooth, continuous spectrum

A perfect absorber/emitter is called a blackbody

Blackbody (thermal) radiation

- A blackbody is an object that absorbs all light, at all wavelengths: perfect absorber
 - No reflected light
- As it absorbs the light, it will heat up
 - Characterized by its temperature
- A black body will emit light at all wavelengths (continuous spectrum): perfect emitter
 - Energy emitted depends on temperature

The radiation the blackbody emits is entirely due to its temperature. Intensity, or brightness, as a function of frequency (or wavelength) is given by Planck's Law:

$$I_{\nu} = \frac{2h\nu^3}{c^2} \left[\frac{1}{e^{h\nu/kT} - 1} \right]$$

where k is Boltzmann constant = $1.38 \times 10^{-23} \text{ J/K}$ Units: J s⁻¹ m⁻² ster ⁻¹ Hz⁻¹

$$I_{\lambda} = \frac{2hc^2}{\lambda^5} \left[\frac{1}{e^{hc/\lambda kT} - 1} \right]$$

Example: 3 blackbody (Planck curves) for 3 different temperatures.



DEMO Hot wire

Wien's law for a blackbody

 $\lambda_{max} = 0.0029/T$

where λ_{max} is the wavelength of maximum emission of the object (in m), and T is the temperature of the object (in K).

=> The hotter the blackbody, the shorter the wavelength of maximum emission

Hotter objects are bluer, cooler objects are redder.



The spectrum of the Sun is *almost* a blackbody curve.





Stefan-Boltzmann law for a blackbody $F = \sigma T^4$

F is the energy flux, in joules per square meter of surface per second, σ is a constant = 5.67 x 10⁻⁸ W m⁻² K⁻⁴, T is the object's temperature (in K).

The hotter the blackbody, the more radiation it gives off at all wavelengths.

radiates more. Visible light Intensity -12,000 K 6000 K 3000 K 3000 500 1000 2000 0

At any wavelength, a hotter body radiates more.

Example:

If the temperature of the Sun were twice what it is now, how much more energy would the Sun produce every second?

See box 5-2 for more examples.

<u>WS1:</u> How hot is the photosphere of the Sun?

Example 2: At what wavelength would the spectrum peak for a star which is 5800/2 = 2900 K?

For a star with T= 5800 x 2 = 11,600 K?

What colors would these stars be?

<u>WS1:</u> How hot is the photosphere of the Sun?

Measure λ_{max} to be about 500 nm, so T_{sun} = 0.0029/ λ_{max} = 0.0029/5.0x10⁻⁷ = 5800 K

At what wavelength would the spectrum peak for star A which is 5800/2 = 2900 K?

For star B with T= 5800 x 2 = 11,600 K?

What colors would these stars be?

Kirchhoff's laws illustrated



Spectrum of the Sun – what kind of spectrum is this?







RA=147.32107, DEC=-0.00658, MJD=51630, Plate= 266, Fiber= 37

Spectrum of a "quasar" – what kind of spectrum is this?

